

Programme Name: B.Sc. in Chemistry (FYUGP)

### PROGRAMME OUTCOMES (POs)

# After completing the Four-Year Undergraduate Programme in Chemistry, Students are expected to achieve the following Programme Outcomes:

#### PO1. Strong grip on fundamental and practical Chemistry knowledge

- Exhibit knowledge of the discipline
- Identify and explain seminal pieces of work in the area
- Conduct guided academic inquiries in various areas of interest in the chosen discipline
- Apply theoretical notions into practice in different forms

#### PO2. Creative and critical thinking, and problem-solving:

- Recognize the social structures underlying our society
- Identify the implications of the same in our existence
- Analyse and engage with their social surroundings, problematize and raise questions based on academic inquiry
- Take informed actions

#### **PO3: Interest in research-based problem:**

- Tools and techniques of research
- •Methodology of research

#### **PO4: Digital Fluency:**

- Develop skill on programming languages
- Effective communication through e-learning

#### **PO5:** Teamwork and communication skills:

• Development of entrepreneurial mind-set, leadership skills and emotional intelligence to face the evolving marketplace.

#### PO6. Professionalism and leadership readiness:

- Act with an informed awareness of issues
- Engage in initiatives that encourage equity and growth for all

## PO, PSO & CO

#### **PO7. Social Responsibility:**

• Function as a collaborating member in teams in multidisciplinary settings

#### PO8. Appreciation and adherence to Ethical integrity:

- Recognize and respect different value systems including one's own
- Follow the norms of academic integrity
- Take cognizance of the moral implications of our decisions

#### PROGRAMME SPECIFIC OUTCOMES

#### Programme Name: B.Sc. in Chemistry (FYUGP)

By the end of the B.Sc. in Chemistry (FYUGP) programme students should be able to:

#### **PSO1: Understand the basics of Chemistry**

- Develop the basic knowledge of chemistry in relation to atomic structure, bonding, periodicity etc.
- Emphasize on different states of matter & their mechanical treatment.

#### **PSO2:** Applications of Chemistry in day to day Life

- Apply the comprehensive understanding of chemical principles to solve practical problems in various fields such as pharmaceuticals, environmental science, and materials science.
- Understand Foundations of chemistry

#### **PSO3:** Development of Analytical and experimental skills

- Analyse the issues to solve them with the design, conduct, and interpretation of experiments
- Enabling students with analysing data critically, and presenting their findings effectively

#### **COURSE OUTCOMES (COs)**

#### Paper title: Core Course -1

#### Paper Code: CHMC1

Nature of Course: CHEMISTRY MAJOR

#### Total Credits: 4

After the end of the course, students will be able to

- **CO1:** Understand the periodic properties of elements, bonding in various molecules.
- **CO2:** Know about properties of gaseous and liquid states of matter, basic organic chemistry and stereochemistry, etc.
- **CO3:** Handle the viscometer for determining the viscosity and different compounds which have immense applications in industry and day to day life.
- **CO4:** Learn the stalagometer for determining the surface tension and different compounds which have immense applications in industry and day-to-day life.
- **CO5:** Purify the various organic compounds through recrystallizations and melting point determinations.

#### Paper title: Fundamentals of Chemistry - 1

Paper Code: MINCHM1

Nature of Course: MINOR COURSE 1

Total Credits: 4

After the end of the course, students will be able to

- **CO1:** Explain the sign of wave function, counter boundary and probability diagrams, different types of bonds and its application.
- CO2: Understand the kinetic molecular model of a gas, behaviour of real gases.
- CO3: Explain the basic organic chemistry and its importance with reaction mechanism
- **CO4:** Analyse the inorganic salt mixture qualitatively.
- **CO5:** Understand the vander waals equation, viscosity of gases.

#### Paper title: Chemistry in Daily Life- I

Paper Code: GECCHM1

Nature of Course: NATURAL SCIENCE

Total Credits: 3

After the end of the course, students will be able to-

- CO1: Understand the composition, processing and analysis of dairy products
- **CO2:** Learn about the various food preservatives.
- CO3: Learn about artificial food colorants
- CO4: Understand about different types of food additive
- CO5: Aware the adverse effects of food adulterants in human health

### Paper title: Core Course -2

Paper Code: CHMC2 Nature of Course: CHEMISTRY MAJOR Total Credits: 4

After the end of the course, students will be able to-

- **CO1:** Understand the preparation, structure and uses of nontransition elements; extraction techniques of metals.
- **CO2:** Understand the various terms and laws of thermodynamics; crystal structure and crystal defects.
- **CO3:** Carry out the preparation & properties of alkanes, alkenes and alkynes etc.
- **CO4:** Estimate iron and oxalic acid indifferent stock solutions provided to the learners which have applications in industry.
- **CO5:** Detect elements and functional groups indifferent organic samples.

#### Paper title: Fundamentals of Chemistry - 2

Paper Code: MINCHM2

Nature of Course: MINOR COURSE Total Credits: **4** 

After the end of the course, students will be able to-

- **CO1:** Understand VBT, CFT and applications of some complexes in various fields.
- **CO2:** Understand VBT, CFT and applications of some complexes in various fields.

## PO, PSO & CO

- **CO3:** Illustrate the applications of solubility and solubility product principle.
- **CO4:** Understand the stereochemistry and conformational analysis.
- **CO5:** Handle pH meter, viscometer and stallagometer for determination of pH, viscosity and surface tension of liquids.

#### Paper title: Chemistry in Daily Life- II

#### Paper Code: GECCHM2

Nature of Course: NATURAL SCIENCE

Total Credits: 3

After the end of the course, students will be able to-

- **CO1:** Understand structure of biomolecules.
- **CO2:** Understand relationship between reactivity of biomolecules and biological functions.
- **CO3:** Demonstrate how structure of biomolecules determines their reactivity and biological functions.
- **CO4:** Explain the various types of vitamins.
- **CO5:** Explain the roles/functions of various types of vitamins.

#### Paper title: Core Course -3

Paper Code: CHMC3

Nature of Course: CHEMISTRY MAJOR

#### Total Credits: 4

After the end of the course, students will be able to-

- **CO1:** Elucidate inorganic reaction mechanism, stability of complexes, substition reactions
- **CO2**: Learn about coordination compounds ligands, theories, d-orbital splitting, electronic spectra, etc
- **CO3:** Understand second and third laws of thermodynamics, ionic equilibria electrolyte, ionization, dissociation
- **CO4:** Learn about cycloalkanes and conformational analysis, chemistry of halogenated alkyl halides
- CO5: Carry out experimental works on inorganic preparation and pH-metry

#### Paper title: Core Course -4

Paper Code: **CHMC4** Nature of Course: CHEMISTRY MAJOR Total Credits: **4** 

#### After the end of the course, students will be able to-

- **CO1:** Understand the principle involved and application of redox reactions.
- **CO2:** Acquire knowledge on acids and bases; understand the chemistry of lanthanide and actinide.
- **CO3:** Learn various laws of electrochemistry, measurements of conductance, applications of electrolysis in industry, electrochemical cells, etc.
- **CO4:** Understand the chemistry of aromatic hydrocarbons and alcohols containing C-O bond.
- **CO5:** Carry out experimental works on conductometry, thermochemistry and qualitative analysis of organic compounds

#### Paper title: Fundamentals of Chemistry - 3

Paper Code: MINCHM3

Nature of Course: MINOR COURSE

Total Credits: 4

After the end of the course, students will be able to-

**CO1:** Learn chemistry of non-metals such as Boron, Silicon. Nitrogen, Phosphorus.

- **CO2:** Understand the general principles of metallurgy.
- CO3: Gain concept of chemical thermodynamics; First and second law of thermodynamics
- **CO4:** Understand the chemistry of Aromatic Hydrocarbons, Alkyl and Aryl halides.
- **CO5:** Carry out Organic Qualitative Analysis practical.

Paper title: Chemistry in Daily Life- III Paper Code: GECCHM3 Nature of Course: NATURAL SCIENCE

Total Credits: 3

After the end of the course, students will be able to-

- **CO1:** Gain knowledge on Chemical and Renewable Energy Sources.
- **CO2:** Understand the basic concept of polymers, classification, characteristics and application of polymers.
- **CO3:** Know the chemistry of Cosmetics & Perfumes.
- **CO4:** Learn different types of fertilizers, manufacture and applications of fertilizers.
- **CO5:** Build strategies for the development of environment friendly polymers.

\*\*\*\*\*\*